

## Absorbance Experiments

Absorbance spectra are a measure of how much light a sample absorbs. For most samples, absorbance relates to concentration:

$$A = -\log \left( \frac{I}{I_0} \right)$$

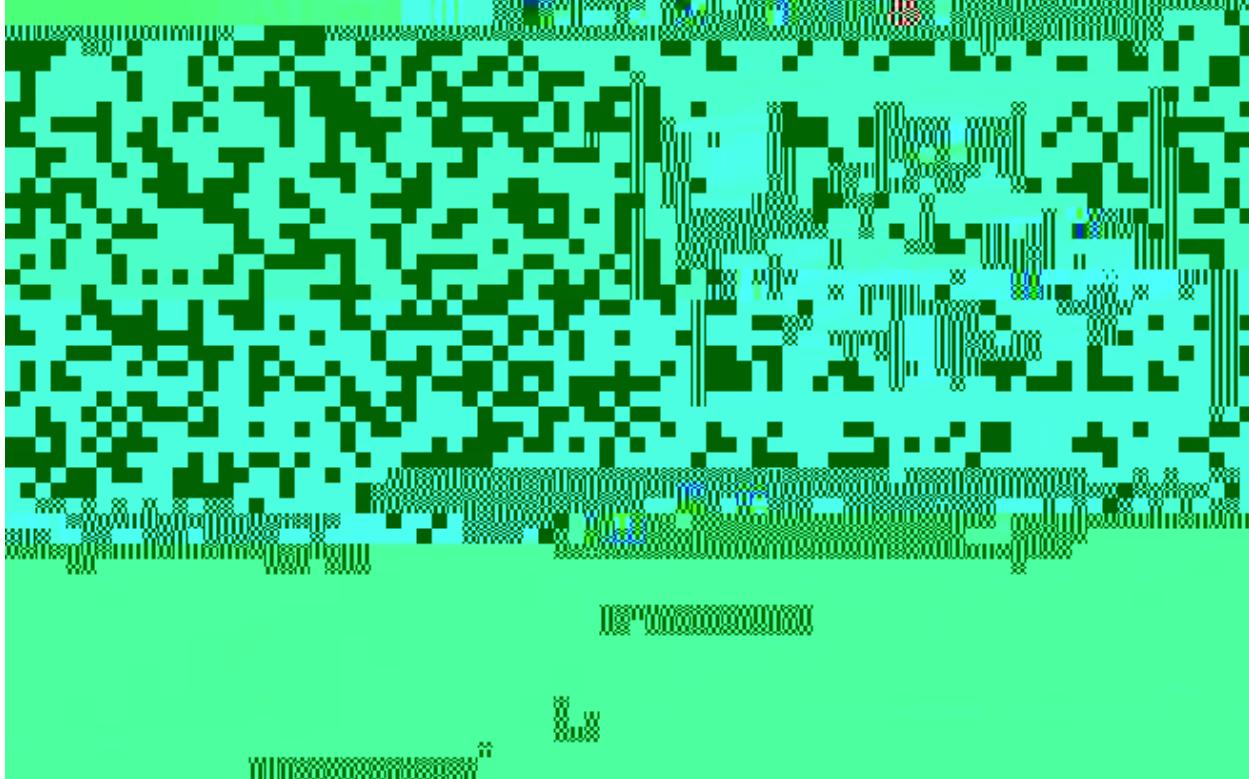
Where:

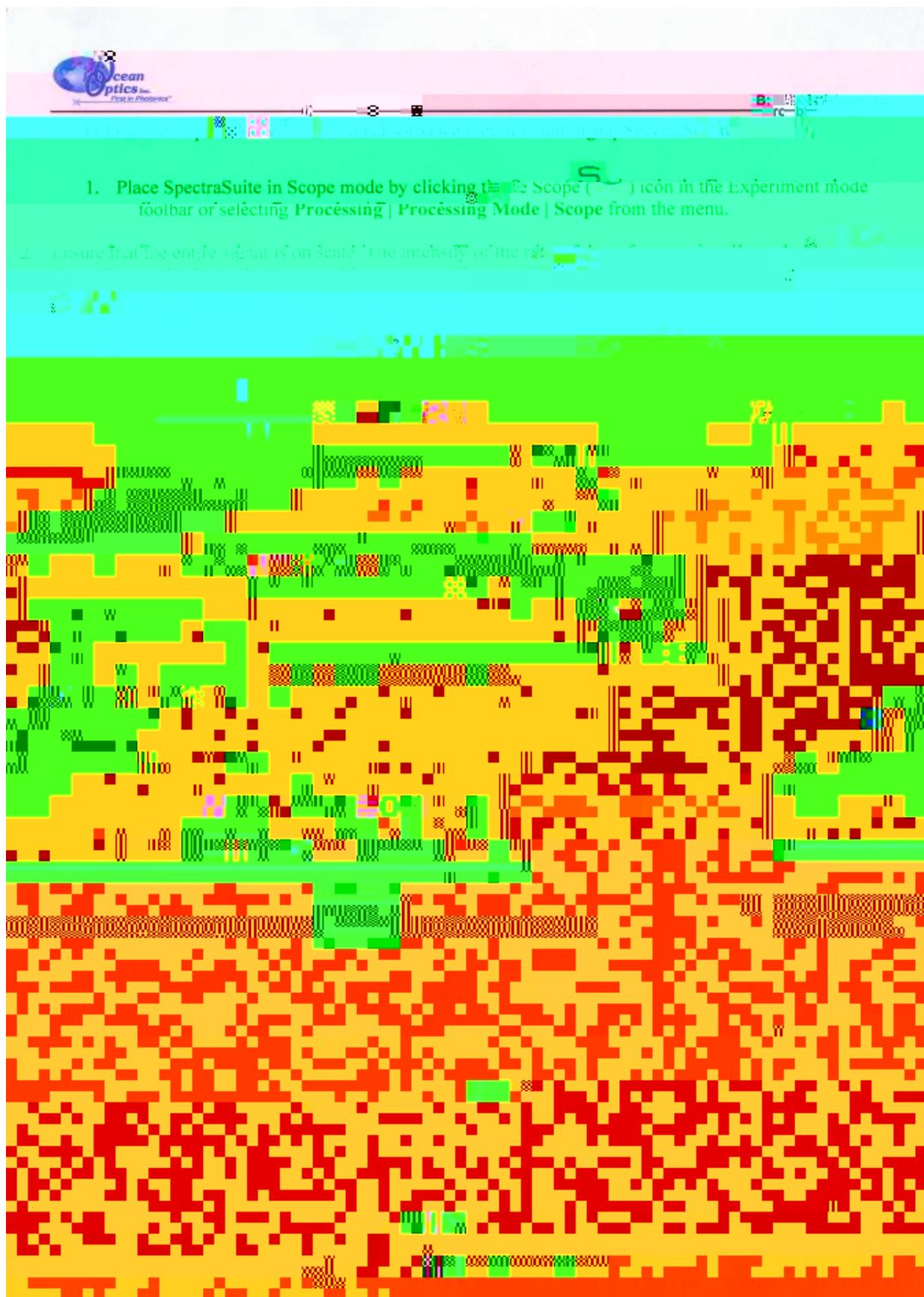
$I_0$  = Sample intensity at wavelength  $\lambda$

$I$  = Intensity at wavelength  $\lambda$

$R_\lambda$  = Reference intensity at wavelength  $\lambda$

Diagram of a spectrophotometer setup. A light source (far right) sends light via an input fiber into a cuvette holder (bottom center). The light interacts with the sample in the cuvette. The light then passes through an output fiber to a detector (top center).









## A: Experiment Tutorials

