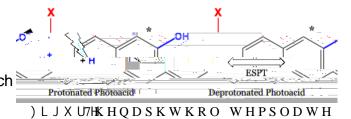
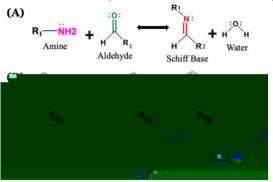
## 7 D N H P D W V X / D E 6 X P P H U 'HYHORSPHQW R I D 6 F K L I I E D V BSK6H, @ WCKLHPW Q R Q3 D0 \$DKWWIKERPD 13RKLR Y

Photoacids, compounds that undergo excited state proton transfer (ESPT), are used to spatially and temporally control pH for several chemical and biological processes, such as acidcatalyzed polymerization and acid-initiated protein folding(Figure



1).<sup>1,2</sup> The Takematsugroup is interested indeveloping synthetic platform that allows for the systematic investigation of structural impacts on the ESPT mechanism and energetics of photoacid candidates. Naphthols, sast-naphthol, are a wet tudied class of hotoacids (Figure 1) Perturbation, snamely D G G L Q J D I X Q F W Ltt to the Draphtto Rstructure can enhance or inhibit photoacidity, create new ESPT pathways, and/or introduce competing excited state pathways. In this work, Schiff basteemistry wasutilized to develop a photoacid library using naphthols as templatesigure 2A) The addition of a Schiff baser carbon nitrogen double bond, introduce photoisomerization, another excited state pathway, which may affect the ability of the photoacid undergo ESP<sup>†</sup>.

From my honors researchlibarary of Schiff base photoacid candidates established

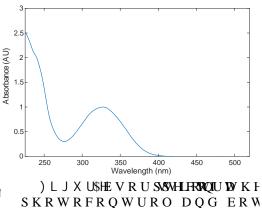


(Figure 2B). Candidate 1 was previously synthesized by the Zeigler groupNuclear magnetic resonance (NMR) spectroscopwasused tocharacterize the Photocontrol and Candidatea2ter synthesisThe purity of theproductswas determined using NMR and steady state fluorescence spectroscopandidate 1 and thePhotocontrol wereound to be pure>98%), but Candidate 2 was crude purification strategy was implemented but was unsuccessful in completely removing unreacted amine and aldehyde starting

) LJXUB\$ JHQHUDO 6FKQWK products I thenutilized ultravioletvisible absorption % VXWHW RI WKH6FKEEPPIW Stody the two photoacid candidates in the ground SKRWRDFLG OLEUL state and their stability in valous solvents (Figure 3).

The project successfully demonstrated that a library of candidates could be prepared

using Schiff base photochemistry. Moving forward, time-resolved absorption and emissispectroscopy will be implemented to study the excitedate pathways of these candidatese Schiff base synthetic platform willalsobe utilized to synthesize new photoacidandphotobase candidateSonducting postbaccalaureatesearch allowed me to advance honors researchandgain experience mentoring undergraduatetudents in lab. The skills I developed this summer will be beneficial as I begin my PhD program at the University of Pennsylvania next year



Citations:(1) Agmon, Noam. J PhyChem A 2005 109, no. 1:353(2) Sambath, Karthik, et al. Org Let 2020 22 3, 1208-1212(3) G.Y. Nages. J of Molec Struc, 2015, Volume 1085.